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Recovery and repurposing of thermal resources in the mining and mineral processing industry

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Abstract

The consumption of energy contributes significantly to the overall cost of operations and the environmental impact of the mining and mineral processing industry. However, despite a few notable exceptions, most of the resulting waste heat produced is dissipated, without recovery, into the environment. There is also a lot of stored heat in mine water which can be tapped into long after a mine has closed. There is, therefore, significant opportunity to improve the industry's sustainability through increasing the amount of waste heat recovered and repurposed.

Keywords

Energy recovery; Mining; Environmental sustainability; Heat pumps; Waste heat

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Recovery and repurposing of thermal resources in the mining and mineral processing industry

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Abstract

The consumption of energy contributes significantly to the overall cost of operations and the environmental impact of the mining and mineral processing industry. However, despite a few notable exceptions, most of the resulting waste heat produced is dissipated, without recovery, into the environment. There is also a lot of stored heat in mine water which can be tapped into long after a mine has closed. There is, therefore, significant opportunity to improve the industry's sustainability through increasing the amount of waste heat recovered and repurposed.

Keywords: energy recovery, mining, environmental sustainability, heat pumps, waste heat

1. Introduction

Industry, as a whole, accounts for around one third of the world's total energy consumption and about 36% of CO₂ emissions [1]. The mining industry is responsible for 4–7% of the total energy consumption [2] and despite an increase in the use of renewable energy, there is still a heavy reliance on fossil fuels [3]. In Canada, for example, CO₂ emissions released from the mining industry grew from about 22.5 Mt in 1990 to 74.5 Mt in 2016 [4]. Energy preservation and management can be accomplished by cutting down on high energy consuming equipment and implementing more efficient technologies [5]. Strides are also being made towards improving energy efficiency through the recovery and reuse of what is otherwise waste heat - the so called 'first fuel' option [6].

It has been estimated that of the total amount of energy consumed in industrial processes, up to 70% is lost as waste heat [7], i.e. thermal energy that is

not captured and is rejected into the environment [8].

Energy usage in mining and mineral processing operations is dominated by a few major processes, including ventilation pumping, comminution and smelting, all of which offer the potential for energy recovery. Mine ventilation, for example, can account for up to 40% of a deep underground mine's total electricity consumption [9], an essential operation to bring fresh air underground and keep areas cool. The smelting process is an extractive metallurgy technique and involves the application of heat to extract metals from their ores [10]. Pumping, particularly in large mining sites, uses significant quantities of energy, which is on average 25%–32% of total motor energy [11]. Comminution (crushing and grinding) reduces the size of the ore to allow further processing to take place. Comminution contributes to approximately 40% of the total energy used in mineral processing [12]. Much of the energy consumed by grinding practices is lost as heat to the ore, resulting in an energy efficiency process of only 1% [11].

Operations, such as process cooling water and off-gasses, expel considerable amounts of unrecovered energy in the form of waste heat [3,8]. Underground

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mining, for example, has several energy intensive stages, including hauling, ventilation, pumping and dewatering [11], with little or none of the generated heat captured. In mineral processing, 20–50% of the primary energy is lost as waste heat [13]; pp. 47–52). It is proposed, therefore, that the industry could significantly benefit from the recovery and repurposing of its waste to offset primary energy purchases in order to enhance overall sustainability.

Whilst advancement in technologies and the improvement of equipment to reach higher efficiencies can limit the release of waste heat and improve industrial sustainable development, studies have shown that one of the most effective ways to improve energy efficiency without the use of major equipment or facility alterations is through waste heat recovery [14]. The recycled energy can, for example, be used to power auxiliary equipment, generate steam, or deliver space heating and cooling [14].

In general, waste heat can be classified as either low or high-grade. High-grade waste heat is at a temperatures greater than 100 °C, and is often used to generate steam or it is recycled directly back into a process [15]. High-grade waste heat can also be captured and used for power generation by driving heat engines, such as those that use the Rankine cycle [8]. Low-grade waste heat is, therefore, typically considered to have a temperature below 100 °C. Temperatures between 70 and 100 °C may be upgraded by adding energy to raise the

temperature to allow for beneficial repurposing, such as steam generation [16,17]. However, at temperatures below 70 °C, currently little of this low-grade waste heat is recovered in the mining industry.

Table 1 summarizes methods of recovering various waste heat sources present in the mining and mineral processing industry, as well as the potential applications for the recovered energy. This review focuses on these various sources of waste heat and the recovery practices and technologies that could be implemented.

2. Underground mining operations

Mining is classified into surface and underground operations, depending on the location, size, depth and grade of the deposit [11]. Surface mining operations do not utilize heating, ventilation and air conditioning (HVAC), which accounts for over 25% of underground mine energy consumption [37]. Underground mining also requires more energy to cope with extended requirements for hauling and dewatering [11]. As a consequence, the focus here is on heat recovery from underground mines.

2.1. Mine water

The need for dewatering in both operating and closed mines presents a promising opportunity for geothermal energy recovery. Mine dewatering usually consists of pumping water from a series of

Table 1. Summary of waste heat recovery methods and potential applications.

Waste Heat Source	Heat Recovered (MW)	Recovery Method	Potential Application	Stage of Recovery Method
Mine ventilation exhaust	9–11 [18,19]	Direct spray recovery with heat pump [19] Heat exchangers [21]	Ventilation preheating [20] Space heating [21]	Potential recovery method
Mine dewatering	1–5 [22]	Heat pumps [23]	Space heating [23,24]	Implemented in industry
Electric Arc Furnace cooling water	10–60 [25]	Heat exchangers [26] Steam generators [26]	Electricity generation [27] Steam [27]	Potential recovery method
Off-gas	60 [28]	Organic Rankine Cycle [29] Bubbled-in off-gas [30]	Electricity generation [29] Microalgae cultivation [30]	Implemented in industry
Slag recovery	9–20 [31,32]	Mechanical crushing with heat pump recovery [33] Centrifugal and air blast granulation using fluidized bed recovery [33] Packed-bed heat exchanger [32]	Preheating air of ore dryers [34] Reheating boiler feed water [34] Electricity or steam generation [35] Combustion air preheating [35]	Potential recovery method
Smelter process cooling water	40 [6]	Heat pumps [6]	Space heating [6] Steam production [36]	Potential recovery method

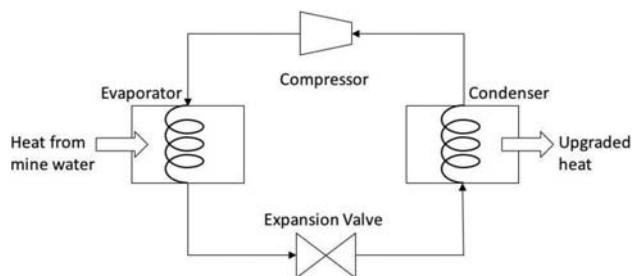


Fig. 1. Heat pump principle in mine dewatering application (adapted from [38]).

wells within the mine to the surface. Dewatering is important, as water levels must be continuously managed to ensure the stability of mine walls and to prevent flooding. Unwanted water can enter a mine from surface accumulations, aquifers, bed separation cavities, solution cavities and old mine workings [23]. Extracting heat from deep underground is beneficial, due to geothermal gradients and large water-rock interfaces resulting in a transfer of heat to the water [22]. The application of geothermal heat pumps can allow for the low-grade thermal energy from warm mine water to be used in a variety of space heating and cooling applications [24].

Heat pumps use external energy (typically electricity) to upgrade heat to a higher temperature. In a compression heat pump, low temperature heat is absorbed in an evaporator and the vapor is then compressed, resulting in the rejection of heat at an elevated temperature in a condenser, illustrated in Fig. 1. Heat is recovered from the condensed fluid, which in turn is expanded and the cycle repeats [5]. The upgraded temperature can be useful for repurposing in applications.

However, a major limitation of the application of the recovered heat is the temperature rise provided by a heat pump, which is typically 25 °C or less [5]. This is seldom exceeded, since compressors with high ejection pressures, which therefore generate higher temperatures, will have increased payback periods due to much higher capital and operating costs [5]. Therefore, with low-grade heat, output temperatures greater than 110 °C are rarely achieved, making the heat recovered by heat pumps unsuitable for many industrial process applications [17].

In an effort to overcome this limitation, chemical heat pump technologies are emerging for the more effective extraction and repurposing of thermal energy. They are driven by reversible chemical reactions and are an environmentally low impact method [39]. The capacity of the system for storage along with the ability to efficiently upgrade thermal

energy and deliver it at high temperatures of more than 200 °C makes it a beneficial solution for low-grade heat recovery [40].

Additionally, after an underground mine is closed, continuous dewatering practices are usually scaled down or ceased, and water levels may rise until the mine is flooded. As a consequence, abandoned mines can still provide a resource for geothermal energy recovery [40,41]. It has been estimated that there are over one million abandoned mines worldwide, with approximately 3000 MW of heat energy remaining untapped in the water within the flooded coalfields of Europe alone [22].

Pumping water from abandoned mines may require a significant amount of electricity, and may be too costly, but there are sites where gravity assists in the discharge of water, or pumping is already required for the maintenance of abandoned mines [22]. In some locations, pumping and treatment facilities are installed to control water levels, as well as treat the water to the extent necessary to make it safe for release to the environment. The treatment technology depends on the characteristics of the mine water, but may include chemical precipitation, membrane applications, and/or biological treatment [42]. As the energy demand of a treatment facility may be very high, particularly due to large pumping requirements, it presents an opportunity for a heat recovery application to off-set costs [22].

Heat pump technology is currently used in many low-grade heat recovery systems for mines. For example, in Great Britain, the escape of contaminated water into the environment from abandoned coal mines is being addressed by the Coal Authority and they have installed 64 treatment systems [22]. A study was undertaken that analyzed 21 of these treatment sites and compared electrical power consumption and thermal power potential [22]. The total low-grade thermal energy of all the mine waters considered was 47.5 MW. Many of the mines that required large power consuming treatment facilities also provided the greatest heat resources due to deep mine waters. For example, for a mine in Durham, England with groundwater temperatures of around 15 °C, it was found that a temperature drop of 4 °C could achieve 4.85 MW of recoverable energy. Whilst at another mine in England, a temperature drop of 4 °C could achieve 806 kWt of recoverable energy. A gravity fed treatment system with water temperatures around 14 °C at Morlais in South Wales was found to offer around 2.90 MW of recoverable thermal energy by reducing the temperature of its water flow by 4 °C [22].

It has been proposed that the application of heat pumps would be a feasible option to capture this

otherwise wasted heat at high efficiencies. In Shetleston, Scotland, a system was installed to recover energy from water in flooded coal mines at a depth of 100 m [23]. Two heat pumps were used to provide space heating by heating the water to 55 °C before storing it in a tank for subsequent distribution to 16 nearby houses. In Heerlen, Netherlands, a geothermal district heating system has been implemented by repurposing thermal energy from several mine shafts in a flooded abandoned coal mine. The system, which provides district heating and cooling to over 350 homes and businesses, was achieved by drilling five wells 700 m deep around the town in order to access underground mine shafts [43]. The wells are able to pump nearly 80 m³ of water per hour at temperatures between 15 °C and 30 °C [44], and a heat pump is used at the surface to extract and generate heat at an average of 28 °C [43].

A heat recovery project in Nova Scotia, Canada was carried out by a plastic packaging manufacturer. It involved the instalment of ground source heat pumps to take advantage of otherwise wasted heat in flood water from abandoned coal mines [45]. At a water temperature of 18 °C, an initial projection found energy savings as high as 70% could be achieved [46]. The water was pumped from 140 m below the surface, extracting heat and lowering the mine water temperature to 13 °C. Eleven heat pumps were installed to provide hot water heating at the plastics plant, at a rated capacity of 985 kWh/day [46]. The capital cost to implement the heat pump system was \$110,000, as opposed to \$70,000 for the alternative propane heating system [46]. However, operating costs of the heat pump system provided annual savings of approximately \$45,000 [47].

A geothermal heat recovery system was implemented in the city of Asturias, Spain, at an abandoned coal mine near the University of Oviedo [48]. The heat from the flooded mine water is recovered using four geothermal heat pumps to provide water heating at a temperature of 46 °C to two of the university buildings, as well as to a nearby hospital. At an average water flow of 215 m³/h, the recovery system reduces the temperature of the mine water from 23 °C to 13.9 °C, allowing for a total annual energy saving of approximately 73% and an annual CO₂ reduction of 39%.

Other types of mines can also be considered, such as the abandoned Wheal Jane tin mine in Cornwall, England, which after flooding began to leak water into the environment. A treatment system was installed to control mine water discharges and involved pumping water from the shaft and treating it with lime. When the treatment facility was

eventually discontinued in 1992, the mine had released 50,000 m³ of water with an average metal concentration of 3500 mg/L to the surrounding environment [23]. It has been suggested that this would be a prime opportunity to install a heat recovery system and continue long-term monitoring and treatment of the mine water [23]. A copper, zinc and sulphur mine in Norway that closed in 1941 established a heat recovery system from the water of the abandoned mine [49]. A heat pump was installed to provide space heating for an underground cavern which is used by the community for various events.

Geothermal heat pumps typically have high initial investment costs due to connections underground, including drilling, piping and excavation [50], costs that can largely be avoided if utilizing an abandoned flooded mine. Con mine, an abandoned gold mine in Yellowknife, Canada is a prospective geothermal heat source for the city based on a preliminary study on the potential to recover low-grade heat for space heating [51]. Of the total energy consumed by the community, approximately 70% is for space heating, resulting in about 277,000 tons of greenhouse gas emissions annually [52]. The study focused on possible technologies to recover heat, as well as approximate costs and payback periods using the example of a 300 kW heat demand [52]. Heat pumps were found to be a potential solution for heat recovery from mine water 400 m below the surface [51]. The temperature would be increased from 35 °C to about 45 °C, with a total power requirement of 90 kW [52]. The net annual savings were projected to be \$95,000 for the project, with a payback period of less than 8 years [52]. The study demonstrates the economic feasibility of heat pumps over time, as they typically have a higher initial capital cost compared to other heating systems, but lower operating costs.

2.2. Mine ventilation exhaust

Through heat transfer from the host rock, geothermal heat is a large contributor to the air temperature of deep mines and the necessary air conditioning and ventilation requirements create a high-energy demand. Exhausted mine ventilation air temperature is typically around 30 °C, with the relative humidity being about 90% [19], and in many cases this heat is released to the atmosphere at temperatures higher than ambient [21]. A ventilations system's wet airflow carries, therefore, a significant quantity of latent heat that could be available for energy recovery and reuse [19]. Waste heat can be extracted from this exhaust air using a medium like water or glycol and then used for

underground ventilation inflow heating or directed to an alternative application, such as space heating of nearby buildings [21].

For mines located in cold climates, heating the air intake can also contribute significantly to the overall cost of operation. The 700 m deep Kylylahti copper mine in Finland, to prevent shafts and equipment from freezing, requires heating the supply ventilation air to approximately 3 °C during months when ambient temperature ranges between 0 °C and –15 °C [20]. The heating system is currently achieved with two gas burners, but with the supply airshaft and the return airshaft only 30 m apart, this provides an opportunity to exchange heat between the two airflows.

A laboratory experiment in China analyzed the combination of a direct spray recovery unit with a heat pump system for the extraction of low-grade energy from air mine exhaust [19]. A multi-stage reverse spray heat exchanger was used, where water is sprayed in the opposite direction of the airflow in multiple stages to improve heat transfer efficiency. This enabled the temperature of the spray water to increase from 7 °C to about 15 °C. A heat pump was subsequently used to extract this heat, with the cooled water returned back to the spray tower. The average exhaust ventilation air temperature and humidity at the surface during cold months are parameters that need to be taken into account to calculate heat recovery potential [21]. A heat pump is an efficient option, since the temperature and humidity of the exhaust air remains fairly constant.

In one of the experiments, 3.47 kg/s of cold water recovered heat at 119 kJ/s from 3.52 kg/s of air when the spray coefficient was 0.9. The results showed that heat exchange efficiency could reach over 85%. This investigation led to the prediction that application in a mine with an air flow rate of about $7.2 \times 10^5 \text{ m}^3/\text{h}$ and using $759 \text{ m}^3/\text{h}$ of cold water could recover about 11 MW.

3. Mineral processing off-gas

3.1. Electric arc furnaces

The production of steel is often accomplished with the use of an electric arc furnace (EAF) [53], technology, which in comparison to a blast furnace, produces less waste and has a lower energy consumption per ton of steel [54]. Electric arc furnaces consume electrical energy and fuels, such as natural gas, to melt and convert solid materials, e.g. iron scraps, into molten steel, which is then further refined to produce high-grade steel [55]. Nevertheless, the iron and steel industry is one of the largest

polluters of the industrial sector [26], contributing approximately 4% of the total global greenhouse gas emissions [54].

The amount of energy required by an EAF to melt scrap iron is 350–370 kWh/tsteel [55] and this releases large amounts of waste heat in an off-gas that is typically around 1250 °C and with flow rates of up to 150,000 Nm³/h [27]. It is estimated that up to 50% of all energy losses and nearly 30% of the energy input is lost in the off-gas from EAFs [53]. Heat recovery from EAFs in the steel industry could, therefore, significantly improve process efficiency, and reduce costs and greenhouse gas emissions.

The Organic Rankine Cycle (ORC) is a piece of technology that can be used to recover high-grade heat from an EAF for use in power generation. In 2013, The ORC production company Turboden successfully implemented a heat recovery system from the off-gas of a 133 t/h EAF used in an iron and steel manufacturing plant in Riesa, Germany [29]. From an off-gas temperature of 1600 °C, the ORC unit used in the design allowed for 3 MW of electrical output, resulting in a payback period of approximately five years.

Off-gases from EAFs must be cooled before they can enter the de-dusting (dust collection) stage. Typically, they are cooled inside a water-cooled hot gas duct to reduce the temperature to the maximum allowable for the dust collectors, about 600 °C, with the removed heat dissipated to the environment [26]. This has been, therefore, an area of particular interest for heat recovery and application research, as the dissipated low-grade energy could be used by a number of process technologies within the steel production plant. For example, waste heat recovered from an EAF could be used for electric energy generation, steam production, or scrap metal pre-heating before the melting process [27].

Energy can be recovered and repurposed at the gas cooling phase through steam generation for use in processes such as shop and office heating, as well as in mills for jacket heating of pickling tanks and fuel oil lines [56]. Steam can also be used for electricity generation using steam turbines [57]. In steam generation, an evaporative cooling system can be installed at the de-dusting system of the EAF and steam generated as water evaporates during the cooling process [53].

A study was performed on a 145-ton EAF [53] to determine the effectiveness of an evaporative cooling system integrated in the cooling water circuit of the de-dusting system at various water pressures. It was found that when the pressure of the boiling water was at an optimal 1 MPa, the amount of steam produced from the heat discharged in the gas was

25.3–28.8 tons per 145 ton batch, with a total exergy of 20–23 GJ.

A waste heat recovery model for steam production from a steel plant was developed in order to find optimized operating parameters [26]. In the model, waste heat was recovered from an EAF used to melt steel scrap and produce molten steel, and the recovery system included heat exchangers, steam generators and a thermocline storage tank. The off-gas was to be cooled to 600 °C from a peak temperature of 1200 °C using cooling water. The cooling water outlet temperature was around 50 °C, but was increased to 200 °C in the model to directly recover heat by producing saturated steam in a shell and tube steam generator. An alternative heat recovery option that utilized heat pumps to upgrade this stream temperature was also considered. The mass flow of cooling water inside the cooling system was controlled using a circulation pump to deliver a minimum mass flow of cooling water in the hot gas line and ensure efficient heat transfer. It was determined from this study that stable and continuous operation was achievable with this approach.

3.2. Smelters

Industrial waste heat recovery from a smelter can provide the necessary conditions required to grow and maintain microalgae year-round in cold regions [30], with the lipids they produce a potential biodiesel feedstock [58]. Due to its relatively clean-burning characteristics, biodiesel can be especially useful in underground mine applications [6,59].

However, the mass production of microalgae is traditionally not possible in temperatures that drop below 15 °C and, therefore, it was previously found to be practical only in year-round warm climates [30,60]. To resolve this issue for cold climates where many mining and mineral processing operations exist, employing waste heat from a nickel smelter site to support algal growth tanks has been investigated [28]. The two major sources of waste heat identified were off-gas streams from the fluidized bed roasters and the furnace. The roaster off-gas undergoes on-site gas cooling from 680 °C to 50 °C to enable the capture of SO₂ through the production of sulfuric acid. Whilst the off-gas from the furnace is released at a temperature of approximately 350 °C [30]. The off-gas streams represent a combined 60 MW of waste heat dissipated into the environment without any recovery [28].

A study modelled the impact on algal pond temperature from direct bubbled-in off-gas from the furnace and heat recovered by a heat exchanger from the roaster off-gas [30]. The results showed

that significant microalgae cultivation could be achieved year-round, even when ambient air temperatures fall well below 0 °C.

4. Slag waste heat

The slag from smelter furnaces represents a substantial amount of “stored” energy, which is lost during solidification. It is estimated that over 40% of the energy supplied to a smelter furnace remains in the slag, with 1–2 GJ of thermal energy contained within one ton [35,61]. Slag has potential commercial value after proper cooling and treatment [62,63], and it is during this process that waste heat recovery can be achieved [64]. For example, blast-furnace slags can be used as a feedstock in the manufacturing of cement, with slag representing approximately 13% of world cement production [65,66].

Traditionally, solidification was achieved through water quenching, with large volumes of water used to rapidly cool molten slags [33,67]. In this technique, molten slag at a temperature of around 1500 °C may be cooled to around 50 °C using a high velocity water stream [67,68]. The water used in this process is recycled and cooled to less than 50 °C, typically by using cooling towers [68]. One of the major drawbacks of this technique is the difficulty of achieving heat recovery using direct application methods, such as heat exchangers, due to the large heat loss through evaporation whilst cooling the slag [69,70]. In addition, low-grade heat is released from the cooling towers to the environment without being recovered [69]. However, this could provide an opportunity for the application of heat pumps to upgrade this low-grade waste heat, allowing for the implementation of a heat recovery system.

With modern slag solidification methods, heat recovery systems that allow for cooling requirements to be achieved are becoming more common. These include the production of steam or hot water for power generation and other industrial processes, or heating the air of ore dryers [71,72]. The recovered energy can also be incorporated back into the furnace process by preheating combustion air [35].

A study using both laboratory-scale and pilot-scale experiments evaluated a heat recovery system using a packed bed of hot slag plates [73]. The molten slag was solidified in the shape of a plate using water-cooled rolls and then it was used in a packed slag chamber where heat exchange took place with a counter-current gas flow. The slag temperature in the pilot-scale test was around 1100 °C, and the maximum gas temperature

measured was 716 °C [73]. From this it was determined that the heat recovery ratio relative to the total heat of the molten slag was 43% [73].

Dry solidification of slag is currently being developed as it offers the potential for an efficient solidification process whilst also providing the opportunity for heat recovery [35]. The process involves breaking up liquid slag into droplets that are passed through a dry cooling stage, where heat can be recovered as they solidify into granules [61]. Using air blast or centrifugal granulation methods in the dry cooling stage, heat is transferred from the droplets to air [33,61,74]. Further heat recovery can be achieved as the solid granules are further air cooled, often with a fluidized bed [35,61].

Early dry granulation techniques involved mechanical crushing with rotating drums [33] and then transferring the broken slag to a cooling chamber where heat exchange could take place with air [33]. However, processing methods such as this typically only allow for low-grade heat to be recovered, wasting a large amount of the thermal content in the slag [34]. Heat recovery from mechanical crushing could, therefore, be an opportunity to apply heat pumps to upgrade the thermal energy to a useful temperature.

New technologies are being investigated to enable recovery at higher temperatures, of which rotating cup granulation is a leading centrifugal method [33,64]. This involves liquid slag being poured into a high-speed rotating cup, which forms droplets that cool as they are ejected outwards. The process produces hot air, typically in the range of 200–300 °C, which is passed through an exchanger to recover the heat and can be used for combustion air preheating or to produce steam or electricity [35,61]. In a commercial trial carried out in Britain, it was found that 59% of the slag heat could be recovered when slag particles were cooled to 250 °C from an initial slag temperature of around 1500 °C [33,61].

Another option is an air blast process, where the molten slag comes into contact with gas traveling at high speed and pressure, resulting in the slag breaking up into small particles [61]. This process again uses a fluidized bed to remove heat from the solid granules. A heat recovery system was adapted to a laterite nickel rotary kiln electric furnace plant in Canada, where slag was granulated using a high velocity air jet inside a waste heat boiler to produce granules [34]. The granules were cooled by radiation to water-cooled boiler walls and by convection to the granulation air. Heat application methods were

Table 2. Comparison of various slag solidification methods.

Slag Solidification Method	Advantages	Disadvantages
Water quenching	<ul style="list-style-type: none"> Produces a glassy slag suitable for the cement industry [32] Efficient cooling of high temperature slag [75] Reduced greenhouse gas emissions compared to cement production from limestone [32] Suitable for large-scale applications [32] 	<ul style="list-style-type: none"> Difficult to achieve heat recovery [69,70] High water consumption [70] Air pollution (SO₂ and H₂S emissions) [75] High energy consumption with additional drying requirements [75] Large particle size [70]
Mechanical crushing with rotating drums	<ul style="list-style-type: none"> Opportunity for low grade heat recovery using heat pumps [34] Up to 60% heat recovery can be achieved [70] Decreases water consumption and gas emissions for minimal environmental impact [35] 	<ul style="list-style-type: none"> Reduced processing capacity due to slag pieces attaching to drum [70] Not suitable for industrial scale operation [32]
Rotating cup granulation centrifugal	<ul style="list-style-type: none"> Minimal environmental impact [32] Small slag particle size [70] 	<ul style="list-style-type: none"> Still in development phase [32] Not well tested for large-scale applications [32] Heat recovery process requires fluidized bed [35]
Air blast	<ul style="list-style-type: none"> Allows for high grade waste heat recovery [35] High glassy phase [32] High process capacity [70] Minimal environmental impact [32] 	<ul style="list-style-type: none"> Heat recovery process requires fluidized bed [35] Large volume of air results in high energy consumption and operating costs [70] Low cooling speed [70]
	<ul style="list-style-type: none"> Allows for high grade waste heat recovery [34] High process capacity [70] 	

established to integrate the recovered energy into the smelter, including preheating air at the ore dryers, reheating boiler feed water, and electricity generation. From this study, it was determined that the heat recovery system could provide up to 25–30% of the total electrical requirements of the rotary kiln electric furnace, as well as reduce CO₂ emissions by 234,000 tons per year [34].

A dry granulation heat recovery technique was developed to recover high-grade heat from molten slag [32]. Slag droplets were formed in a dry granulator process, which were cooled to produce glassy granules at a temperature of about 900 °C. These granules were transferred to a packed-bed heat exchanger to allow for heat recovery from the slag. It was found that the system could allow for approximately 80% of the heat to be recovered to produce hot air or steam at temperatures above 600 °C, which could be used in drying and preheating applications.

The challenges associated with employing various slag solidification methods are demonstrated in Table 2.

5. Process water

During the study of a smelter, a number of process streams in the facility were identified for low-grade heat recovery [6]. The smelter processed mainly nickel and copper custom feeds and concentrates, with annual production of around 75,000 tons of nickel and 23,000 tons of copper. With nickel and copper bearing sulphide ores as the primary feedstock, the total waste heat rejected to the environment from the fluidized bed roaster off-gas, the furnace off-gas, the furnace cooling water, the matte granulation cooling water and the acid plant cooling water during operation was calculated to be in excess of 100 MW [6].

The most promising streams were found to be process cooling waters from the calciner, the matte granulation process and the furnace, as these were all in the temperature range (20–90 °C) which is suitable for the application of heat pumps [76]. It was determined that for the beneficial application of the recovered energy, the waste heat would first need to be upgraded by heat pumps. This offered the potential to provide space heating and cooling of a nearby office building, and to replace an aging natural gas furnace with an average annual demand of 510 kW, as well as reduce CO₂ emissions by approximately 62% [6].

An iron and steel plant located in the Netherlands was investigated to determine the feasibility of implementing a heat recovery system using cooling

water from the rolling process, where rolled steel is cooled by spraying water at a temperature of 80 °C [36]. A preliminary study found that waste heat from the cooling water of the hot strip mill can be recovered using an absorption heat pump, which could then be applied to produce low pressure steam at 1.7–3.5 bar and 130 °C [36]. It was determined that emission reductions of 1.9 kg CO₂/ton of steel could be achieved with the implementation of this technology [36].

6. Conclusion

The mining and mineral processing industry is extremely energy intensive, yet in general carries out relatively little waste heat recovery and repurposing. Many sites are located in very cold climates where recovery and repurposing on-site to supply space heating or the pre-heating of gas and liquid process streams could significantly contribute to the sustainability of the operation. Where recovery does exist, the target is generally high-grade waste heat sources in gasses or liquids with temperatures greater than 100 °C, most of which is related to smelter furnaces. This high-grade heat in the exhaust gas from electric arc furnaces, for example, can be extracted for direct recovery applications, such as steam generation, electricity production, or process stream preheating. In most cases use of heat exchangers, which are well understood and widely used throughout all industries, prove to be the most appropriate recovery route. Although, over 40% of the energy supplied to a smelter furnace remains in the slag, representing substantial energy that is rarely tapped. High-grade waste heat can also be captured and used for power generation by driving heat engine technologies such as the Rankine cycle.

Significant opportunity also lies in the abundant quantities of low-grade thermal resources which are less than 100 °C. Recovered energy, again usually with heat exchangers, from process streams with temperatures between 70 and 100 °C may be upgraded by adding energy to raise the temperature to allow for repurposing, e.g. through steam generation. However, the copious quantities of resources that are below 70 °C are usually discounted by the industry as too difficult to capture and repurpose. However, it is proposed that an effective way to recover this low-grade waste heat, is upgrading it with heat pumps. This can allow for the recovery of waste-heat as an economically viable option as well as reduce overall greenhouse gas emissions. Geothermal heat pumps, which can provide a lift in temperature from 15 to 20 °C to around 30–40 °C can be effective for the recovery of

waste heat within flooded abandoned mines for use in local space heating, such as in nearby homes or community centers. For operational sites, such as smelters, process cooling streams offer the most potential and chemical heat pumps are the more appropriate option as they can provide a temperature increase greater than 25 °C. The recovered heat becomes more suitable for displacing existing process stream heaters or to be used to replace fossil fuel fired boilers to supply space heating, thereby reducing both annual operating costs, as well as CO₂ emissions.

Conflict of interest

None declared.

Ethical statement

Authors state that the research was conducted according to ethical standards.

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